

# The Use of Urinary Porphyrins Analysis in Autism

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## Abstract

The analysis of urinary porphyrins promises to provide substantial insight into treating autism. Previous methods to determine body burden of toxic metals relied on a chelator challenge. However, the urine porphyrins test is more effective than a chelator challenge in determining toxic metal body burden because it measures the effects of those metals on the body. Specifically, certain metals, particularly toxic metals such as mercury, lead, and arsenic, will inhibit different enzymes of the heme porphyrin pathway and will thus cause different and specific porphyrin patterns (or “profiles”) in the urine, the analysis of which can help determine which metal is involved, and to what degree. The higher the toxic metal exposure and body burden, the higher the porphyrins are elevated in the urine. In humans, urinary porphyrin profiles directly reflect mercury body burden and neurobehavioral deficits.

This article reviews one recent prospective study of 115 children with autism who demonstrated porphyrinuria when compared to 119 control children, including a mean increase of 2.6-fold ( $p < 0.001$ ) in coproporphyrin. The elevation in coproporphyrin also correlated with the severity of autism and was consistent with mercury exposure and burden. A subgroup of these autistic children underwent oral chelation therapy with meso-2,3-dimercaptosuccinic acid (DMSA) which resulted in a significant reduction in mean urinary coproporphyrin and precoproporphyrin ( $p = 0.002$ ). Another prospective study on 37 autistic patients confirmed that the severity of autism was directly correlated to the degree of porphyrinuria. The interpretation of the urinary porphyrin profile is reviewed, as are important differences found in males versus females.

This paper is based on a transcript of one of my Autism One radio shows (see <http://www.autismone.org/radio/?archive=1613>), *Autism Innovations*, which was broadcasted on January 4, 2007. It was modified to enhance readability and to provide references to the relevant medical literature. This paper reflects my current opinion and understanding of the medical literature, but might not be representative of the views of other physicians. Talk with your physician about any questions you might have.

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## Introduction

The analysis of urinary porphyrins promises to provide substantial insight into treating autism. Before diving into porphyrins, let’s start with a quick overview of what happens when toxic metals, such as mercury and lead, arrive in the body. When we are exposed to mercury or lead, whatever is not immediately excreted will circulate in our blood for about a month or so [1] and eventually will be deposited deep in body tissues. Mercury is mainly deposited in the kidney, liver, and brain. Lead is typically deposited in the brain and bone. So if you are poisoned by mercury or lead during a single acute exposure event, and then measure a blood or urinary level several months later, those levels will be essentially zero [1].

Prior to the advent of urinary porphyrins analysis, we would perform a chelator challenge where we would administer to the patient a chelator that has a negative charge, and it would find things with positive charges, including metals such as mercury and lead, bind them, and then the body would excrete those bound metals in the urine. We would then measure how much of the bound metals appeared in the urine. However, if the metals were located deep in body tissues, then the chelator typically would not pick them up easily and thus we would not see many metals in the urine. If we saw a lot of metals in the urine after the chelator challenge then that was great; however, often we did not. So the chelator challenge usually did not let us

know what the body burden of metals was in a particular individual, and sometimes missed certain metals completely.

In my opinion, the urine porphyrins test is more effective than a chelator challenge in determining toxic metal body burden because it measures the effects of those metals on the body. Here is why.

Porphyrins are a special class of uniquely-structured and uniquely-sized compounds that can surround and bind nearly any metal ion, and thus have diverse biological functions (see, e.g., <http://en.wikipedia.org/wiki/Porphyrin> or <http://www.answers.com/topic/porphyrin>). For example, heme proteins (which contain iron porphyrins) are very common in nature and serve many roles, including binding, transport, and storage of oxygen (myoglobin and hemoglobin), electron transport (cytochromes b and c), oxygen activation and utilization (cytochrome oxidase), and managing and using hydrogen peroxide (peroxidases and catalases). Other porphyrins play a role in detoxifying certain chemicals such as pesticides and other toxins (P450 enzymes). Related porphyrins include the chlorophylls (which have a central magnesium ion) and pheophytins (which are metal free), which are respectively used by plants and bacteria for photosynthesis. Another common porphyrin is vitamin B-12, which has a cobalt ion, and which is found in both bacteria and animals. Zinc, copper, and nickel containing porphyrins are also known, and many other metals can be inserted in porphyrins.

Porphyryns are involved in the production of heme which is a necessary ingredient of the P450 enzymes which are critical for detoxification of chemicals and toxins including pesticides. Heme is also necessary to remove beta-amyloid plaques from the brain [2]. In a recent study, children with severe autism secreted 2-3 times more beta-amyloid precursor protein than children without autism [3]. To generate heme, via specific enzymes, the body normally converts one species of heme porphyrin to another, in a sequential metabolic order or pathway. Yet, certain metals, particularly toxic metals such as mercury, lead, and arsenic, will inhibit different enzymes of the heme porphyrin pathway and will thus cause different and specific porphyrin patterns (or “profiles”) in the urine, the analysis of which can help us determine which metal is involved, and to what degree [4]. For example, biphenyls, dioxins, and aluminum inhibit uroporphyrin decarboxylase, which is an early enzyme in the porphyrin pathway, while mercury inhibits coproporphyrinogen oxidase, which is a later enzyme, and lead inhibits both copro-porphyrinogen oxidase and aminolevulinic acid dehydratase.

In short, if a heme porphyrin enzyme is inhibited, the corresponding porphyrin will not be able to convert to the next in the metabolic pathway, as would normally occur if the enzyme were uninhibited. Thus, that affected porphyrin will build up in the body, and be excreted in the urine. The more types and/or quantities of inhibiting metals that are present in the body, the higher the inhibition and the more types and/or quantities of affected heme porphyrins will appear in the urine (and in the analysis of that urine, which we call the “urinary porphyrin profile”). Wherever the inhibiting metal is located in the body, whether deep in the tissue or superficial, the porphyrin-converting enzyme inhibition will likely take place and the corresponding porphyrin will appear in the urine. Consequently, by measuring the effects of these particular metals in the porphyrin pathway, we can indirectly determine which metals are present in the body, and generally in what quantities.

### Urinary porphyrins studies

The study of porphyrins in human diseases was pioneered by Dr. Jim Woods out of Seattle, who began this work roughly 30 years ago. At the Seattle 2006 Fall DAN! Conference, Dr. Bradstreet and I met with Dr. Woods, and he spoke during Dr. Bradstreet’s lecture. At ICDRC, we are currently collaborating on some studies with him, so stay tuned as we think the results will be exciting and surprising to many people.

Many studies have demonstrated a link between exposures to toxic substances and porphyrin problems. For example, a study by Fowler showed that the metabolic pathway for the heme porphyrin is “highly sensitive to inhibition” by metals such as lead, mercury, and arsenic [5]. In animal studies, Miller demonstrated that mercury exposure causes an excretion of excessive porphyrins in the urine (called porphyrinuria), including elevations in urinary excretion of pentacarboxyporphyrin, coproporphyrin, and precopro-porphyrin, compared to unexposed control animals [6]. Woods demonstrated in a rat study that porphyrinuria caused by methylmercury first occurred at doses that did not produce

obvious neurological problems [7]. Gonzalez-Ramirez later replicated these findings in humans exposed to mercury vapor [8]. Heyer showed that the urinary concentration of porphyrins can serve as “a biomarker of prolonged exposure to all forms of mercury.” Furthermore, the higher the mercury exposure, the higher the porphyrins were elevated in the urine [9]. Woods demonstrated that in humans, urinary porphyrin profiles directly reflect mercury body burden and neurobehavioral deficits [Woods, 1996]. Echeverria found declines in “neurobehavioral performance” at mercury concentrations less than 4 µg/l in the urine, which is an extremely low level and approaches what is found in the general population [10].

The key to the urinary porphyrin test is that it can help determine the body burden of toxic metals, which can otherwise be problematic to determine. Previous studies demonstrate that some chronic symptoms (present more than 1 year) caused by mercury can remain undiscovered if evaluations rely only on prechelation urinary measures of mercury exposure [11]. This is because urinary measures of mercury prior to chelation reflect only recent exposure (over the previous 2-3 months) [9]. In other words, the amount of mercury present in the urine before administration of a chelator does not correlate with the body burden of mercury. However, the amount of mercury present in the urine after a chelation challenge with dimercaptopropanesulfonate (DMPS) is a measure of mercury exposure [8] and estimates the body burden of mercury [11,12]. Animal studies demonstrate that treatment with DMPS significantly decreases the kidney concentration of mercury and increases urinary mercury excretion. In addition, DMPS treatment decreases both kidney and urinary porphyrin concentrations, which are highly correlated ( $r \sim 0.9$ ) with the decrease in mercury concentration in the kidney. Therefore, urinary porphyrins are a biomarker of the renal mercury content [12]. In addition, if autistic children have an increased body burden of toxic metals compared to control children, we would expect more toxic metal urinary excretion after a chelator dose [10].

### Genetics and porphyrins

Several studies demonstrate a link between specific genetic variations and certain porphyrin problems. For example, Woods recently described a polymorphism in exon 4 (N272H) of the gene that encodes for the production of the heme pathway enzyme called coproporphyrinogen oxidase (also known as CPOX-4). The homozygous genotype occurs in approximately 3% of the population and the authors hypothesized that a CPOX-4 polymorphism might increase the risk of neurotoxicity upon exposure to mercury [13]. In another study by Heyer, people who were homozygous recessive for the CPOX-4 polymorphism had evidence of increased neurotoxicity to dental mercury exposure when compared to the remaining population. In other words, the study was “suggestive that the CPOX4 polymorphism could increase susceptibility to mercury-related neurobehavioral effects” and therefore the CPOX4 polymorphism “might serve as a biomarker of susceptibility to mercury toxicity” [9]. This supports the idea of a genetic susceptibility to mercury toxicity, something that many of us have stated about autism for some time now [14, 15]. Heyer and Echeverria also demonstrated a genetic

susceptibility to mercury toxicity in humans exposed to chronic low-level mercury in which a polymorphism in brain-derived neurotrophic factor (BDNF) was associated with increased self-reported mood and symptom scores ( $p = 0.0039$ ) [16] and changes in cognitive and motor function [11]. In other words, people who are homozygous recessive for the CPOX-4 and BDNF genes appear to be more susceptible to mercury toxicity compared to the general population. Interestingly, problems in BDNF are present in some autistic individuals [17]. Further study is needed to determine if children with autism are more likely to have polymorphisms in these 2 genes compared to the general population. Since Woods showed that the rate of CPOX-4 polymorphism is approximately 3% in the general population [13] and Echeverria showed that the BDNF polymorphism occurs in approximately 4-5% [11], large scale population studies might miss increased susceptibility to mercury toxicity in this small portion of people. Any problems found in 1 or 2 people during large studies tend to be averaged out because of the large number of people involved (in studies these people are usually called “outliers” and are excluded from analysis). The CPOX-4 gene defect is present in about 3% of population, but might be higher in autism. As an aside, Woods has shown that people with low glutathione levels have more porphyrins in the urine [18] because there is less glutathione available to bind to the metals and prevent toxicity. Of course, we know that most children with autism have documented lower levels of glutathione [19].

### Porphyria and autism

One recent prospective study of 115 children with autism demonstrated porphyria when compared to 119 control children [20]. In this study, 53% of the autistic children had elevated urinary porphyrin levels. This study enrolled 115 children with autism, 63 with PDD-NOS, and 88 children with other neurological disorders such as Asperger disorder, attention deficit, hyperactivity, epilepsy, and cerebral palsy. When compared to the control group, children with autism had a mean increase of 2.6-fold ( $p < 0.001$ ) in coproporphyrin (which I refer to as “copro”). This elevation persisted when normalized to urinary levels of creatinine. The elevation in coproporphyrin also correlated with the severity of autism. The autistic children also had significantly higher levels of mean urinary precoproporphyrin (“preco”,  $p < 0.001$ ), pentacarboxyporphyrin (“penta”,  $p < 0.001$ ), and hexacarboxyporphyrin (“hexa”,  $p < 0.002$ ). There was also a slight elevation of uroporphyrin levels in the children with autism which could be consistent with xenobiotic (chemicals or pesticides) exposure, but the difference was not significant versus the control group. A subgroup of these autistic children underwent oral chelation therapy with meso-2,3-dimercaptosuccinic acid (DMSA) which resulted in a significant reduction in mean urinary coproporphyrin and precoproporphyrin ( $p = 0.002$ ), indicating that the urinary porphyrin elevation was not genetic in nature but due to the toxic metals removed. Children with autism and seizures had even higher levels of mean coproporphyrin compared to children with autism without seizures. This is not surprising because seizures can be induced by certain toxic metals [21],

especially mercury [22], and Dr. Nataf showed that mean copro levels in children with seizures were higher than in children with autism [20].

Another prospective study on 37 autistic patients confirmed that the severity of autism was directly correlated to the degree of porphyria. Non-chelated autistic children had more porphyrins present in the urine compared to children with PDD, Asperger syndrome, or control children. Control children were neurotypical and matched according to age, sex, and race. Non-chelated autistic children also had higher levels of mean coproporphyrin (1.7-fold) when compared to chelated autistic children ( $p < 0.05$ ), and almost 3-fold higher when compared to the control group ( $p < 0.05$ ) [23]. This is also what we are finding clinically at ICRDC—in general, the more severe the autism in a child, the higher the urinary porphyrin levels.

In some studies, elevated porphyrins appear to directly cause toxicity. Ruscito showed that some porphyrins interact with the binding site for benzodiazepines on the gamma-aminobutyric acid (GABA<sub>A</sub>) receptor and causes synaptic transmission inhibition, much like diazepam [24]. Also, some porphyrins, if the levels are high enough, appear to directly cause neurotoxicity [25].

### Applying the urinary porphyrins test

So how do I use the porphyrin test in autistic children? The history is always the most important place to begin. I first look at the metal exposure history of the child. Since 1 out of 6 children in the United States is born with blood mercury levels high enough to impair neurological development [26], I can expect that some children will have this as a main exposure. The amount of fish mom ate and the number of maternal amalgams present during pregnancy are important because both of these correlate with the amount of mercury present in the fetal blood and brain in utero [27]. In many studies, the fetal blood level of mercury was shown to be approximately twice that of maternal blood [28, 29]. Furthermore, many children are exposed to lead during pregnancy as maternal bone is broken down to release calcium and thus also lead [30]. Other exposure history in some children includes immunizations containing thimerosal, fish consumption, lead in the environment which is practically ubiquitous, antimony from flame retardant bedding and pajamas, arsenic from chicken and treated wood, pesticide exposure, etc.

So, before ordering a urinary porphyrin profile, I have an idea of the exposure history and what the porphyrin test might show. However, I don't know what amount of the toxic metals each child is going to retain after exposure. What we think is occurring in autism is an inability to excrete certain metals and toxins [14, 15]. In many children, I have a general idea that the porphyrin profile will be elevated but not how high it will be. However, sometimes I am surprised by the profile or wrong, so ordering the profile is important. I do find that the severity of elevation on the porphyrin test helps me to know how aggressive to be with chelation.

As an aside, if your child has a fairly high toxic metal exposure history but a porphyrin profile that is very low, then it may be a false-negative. A false negative can occur if the porphyrins become oxidized by substances or damaged by light.

Oxidation of porphyrins can occur from supplements that are excreted in the urine and many of our children are on high doses of nutritional supplements. Therefore, a second morning urine is more helpful in that case, as supplements are concentrated in the urine throughout the night. Dr. Woods and his colleagues have also found that a second morning urine is more accurate in general [31], and this is usually what we recommend. However, a second morning urine can also lead to a low creatinine level in the urine which can affect the interpretation of the lab. Dr. Nataf has indicated to us that the porphyrin test is accurate as long as he can run the sample within about 3 weeks and it is protected from light. I generally recommend collecting the porphyrins in a glass or paper vessel if possible to minimize contamination from other types of containers. The sample can be shipped to France by U.S. postal mail, costing typically less than two dollars.

False negative results appear to be more common in women. In some studies, the excretion of porphyrins in the urine is sporadic in women but not as much in men (we assume the same holds true for children, but these studies were done in adults). The excretion of porphyrins in most people has a diurnal variation [6], which means that it has peaks and valleys throughout the day. In men the excretion of porphyrins in the urine is fairly constant, and Woods has shown that a spot urine correlates well with 24 hour urine porphyrin levels in men. However, in women, the excretion of porphyrins is more variable throughout the day and therefore a spot urine does not always correlate well with 24 hour urine porphyrin levels [32]. In girls, we still do a spot urine test (mainly because a 24 hour urine test is not always feasible), but if it the porphyrin levels are low (especially if exposure levels are high), then we strongly consider repeating the porphyrin sample or collecting a 24 hour urine porphyrin sample if possible. Since porphyrin excretion varies throughout the day in most people, when repeating porphyrins to assess treatment outcomes, it is important to collect the urine at the same time of day as the previous sample in order to make comparisons more reliable. One study demonstrated that the best time to collect a urine sample for porphyrins was mid-morning as this correlated best with 24 hour urinary porphyrin levels [31]. Therefore, I generally recommend performing a second morning urine collection.

Porphyrin excretion is representative of the toxic metals present in the kidney. Therefore, it does not necessarily represent what metals are present in the brain or in what quantities. Currently, short of a brain biopsy, we do not have a lab measurement for types or amounts of toxic metals that are present in the brain. However, we assume that the amount of toxic metals present in the kidney are reflective of what is present elsewhere in the body, including the brain. In multiple studies, the amount of porphyrins in the urine correlates with the amount of metals present in the kidney upon autopsy. Furthermore, as urinary porphyrin levels decrease with chelation, the amount of mercury present in the kidney drops in proportion to the decrease in urinary porphyrins [12]

So which lab should you use to measure urinary porphyrins? I prefer and use Dr. Nataf's lab in France. His group was the first to describe the use of porphyrins in the urine of autistic children. Based upon his study, we know the mean values of

urinary porphyrins in control children and in children with autism, PDD, Asperger syndrome, seizure disorder, etc. These values are helpful because they provide reference ranges that we can use in interpreting the test results. I tend not to use LabCorp, Quest, and other U.S. labs because the norms for these labs are typically for adult healthy patients and not for autistic children. Furthermore, many of the U.S. labs use the porphyrin test to rule out genetic porphyrias where massive amounts of porphyrins are present in the urine. Thus they can sometimes miss the small elevations that we find in autistic children because their lab machinery is not calibrated properly. In addition, the units of measurement used in Nataf's lab and the U.S. labs are different, and need to be converted to nanomoles per gram of urinary creatinine (Nataf's unit of measure) to compare measured values to the reference range in autistic children as described by Dr. Nataf (this can be quite tedious). Furthermore, Dr. Nataf's lab and Dr. Wood's lab in Seattle are the only labs that I know of in the world that can measure preco levels (Dr. Wood's lab only does it for research purposes). This marker is extremely important as discussed below.

When interpreting urinary porphyrin levels, I first look at the amount of creatinine present in the urine. This is typically listed on the very bottom of the first page of the Nataf report. I like to see around 1000 mg/l. If it is less than 1000 mg/l, then the amount of porphyrins reported as present in the urine could be falsely elevated. If it is more than 1000 mg/l, then the amount of porphyrins reported could be underestimated. However, this is not a hard and fast rule because children with autism often have lower creatinine levels than neurotypical children [33], which makes interpretation of all this somewhat tricky. Suffice it to say that if the urinary creatinine is near 1000 mg/l plus or minus 200-300 mg/l or so, I am generally happy with that. If the urinary creatinine is out of this range, then we have to be careful with the interpretation, and sometimes it is worth repeating the sample. However, often times the porphyrin values are so elevated, that even if the urine is too dilute and the creatinine is low, the urinary porphyrin values are still quite elevated even after correction.

Next, I look at the copro level. Typically this is elevated with respect to the listed reference range due to exposure to mercury, lead, and/or arsenic. The copro level is a good marker of *overall* toxic metal burden in the kidney. Then, I look at the preco level. Since the only thing that is known to elevate the preco value is mercury [13], preco is a marker of mercury exposure and nothing else. Preco is called an atypical urinary porphyrin because it is not normally present in the urine, and is only found with elevated mercury levels in the body. If the copro and preco are elevated by about the same percentage over their respective reference ranges, this is usually a good reflection of mercury. If the copro is more elevated in comparison to the percentage of elevation in preco, then usually another metal is heavily at play here, typically lead or arsenic. If the percentage of increase in preco is more than the percentage increase in copro, then mercury is predominant. Next, I look at penta which is elevated with mercury exposure, but is typically less elevated than preco because it is upstream of the enzyme inhibited by mercury in the porphyrin pathway. Finally, I look at the heptacarboxyporphyrin ("hepta") and uroporphyrins

(“uro”) levels. These are typically elevated over their reference ranges due to xenobiotic exposure such as PCB’s, pesticides, and other toxic chemicals, such as dioxins. However, arsenic and aluminum can also cause these to be elevated. So with elevated uroporphyrins, it can be difficult to tell if the elevation is from toxic chemicals, arsenic, and/or aluminum.

For Dr. Nataf’s lab, he also lists several ratios that can be helpful in interpreting the lab results, along with respective reference ranges for each ratio. For example, a preco/uro ratio is provided. Since uro is not elevated by mercury, but preco is, the ratio of these two is helpful to confirm that the preco elevation is real and not a false positive result. Also listed (sometimes on the second page) is a preco/copro ratio. An elevated preco/copro ratio can tell us that not only is mercury present, but that it is the predominant metal. A preco/copro ratio that is lower than the reference range (but copro is still elevated) indicates that another toxic metal like lead or arsenic is predominant. The final ratio listed is an uro/copro ratio. An elevated ratio of uro/copro is consistent with xenobiotics, arsenic, or aluminum exposure and confirms that the elevation in uro is not a false positive. As an aside, we know that some children with autism have genetic defects in the paraoxonase gene and therefore cannot properly detoxify pesticides [34]. It must be noted that many times, the children we evaluate were exposed to many different toxic metals and chemicals, including mercury, lead, aluminum, arsenic, pesticides, PCB’s, and others. Therefore, the ratios can be somewhat tricky to interpret as we have to figure out how each metal or toxin will affect the overall porphyrin profile.

Interpreting mercury exposure on the porphyrin test is relatively easy by looking at the preco level. However, lead is a little more difficult to interpret. Generally with lead exposure, it will be reflected by a high copro level, but this can also be from mercury or arsenic. An exposure pattern which is consistent with lead only and not mercury (which I see on occasion) is a high copro and a much lower preco level. It is my opinion that it is generally important to measure a blood lead level to exclude active exposure to lead, as lead poisoning is known to mimic autism [35]. In fact, I have discovered a few children who were getting active lead exposure and this led to their autistic symptoms that improved when the lead exposure was removed. Some unusual lead exposures that I have seen are from physical therapy vests made of lead or old bathtubs with cracks in them that release lead. Arsenic and aluminum burdens are very difficult to interpret based upon the porphyrins test because the results overlap with other toxins. As an aside, I have started receiving some porphyrin results on parents of autistic children, and finding high porphyrin levels in some.

Once a high porphyrin level is established, I find it helpful to then sometimes perform a chelator challenge and obtain a provoked urine toxic metal test if the type of metal present is still in question. Using different chelators can sometimes be helpful to know how a child will respond to each. I prefer suppositories or IV’s as I generally see the best results with these. In addition, I am doing a lot less chelator challenge tests now because of the porphyrin test. If the preco is fairly elevated, then I know I need to use a chelator that will remove mercury. So typically I will start with DMSA or DMPS, but I do also use a lot of calcium-disodium-ethylenediamine-

tetraacetate acid (EDTA), as many of the children I am seeing have evidence of unrecognized elevated lead levels. Sometimes this means I am using DMPS and EDTA together. Of course, the choice of chelator(s) is individualized to each child. Recently, some people have expressed concern that sulfa-based chelators may not be tolerated by some autistic children based upon genomic testing. However, based upon my review of the literature, a very consistent finding in most children with autism is a deficiency in sulphates which can lead to a decreased ability to properly metabolize toxins [36]. If an increase in sulphates is suspected based upon genomics, it is quite easy to measure a blood sulphate level and a urinary sulphate level and determine if there is truly an increase. If a child has very high urinary porphyrin levels (i.e. copro > 400 or preco > 25), it is my opinion that removing the toxic metals without using DMPS or DMSA is going to be difficult. Furthermore, an IV or suppository form of chelation is sometimes required.

We might be able to use the porphyrin test to determine when to stop chelation. However, a recent study on using chelation in rats to remove lead from the brain showed that lead was still being removed from the brain even after the measured blood lead level declined to zero [37]. Therefore, this *might* mean that we want to continue chelation for a period of time even after the porphyrin levels or urine toxic metal tests drop to normal. Nataf was able to use DMSA chelation orally to bring the porphyrin levels to normal [20], so we might be able to use the urinary porphyrin test sequentially to help us determine when we are nearing the end of chelation. However, we are still in the process of evaluating this. In general, for most children, I can say that the urinary porphyrin levels are coming down with chelation over time, and autistic symptoms in many children are improving.

So, is the urinary porphyrin test reliable? Dr. Woods has done research on this test for about 30 years and has shown that it is very reliable, especially for mercury exposure. Recent studies have shown the amounts of porphyrins present in the urine prior to commencing chelation correlate well with the severity of the autism [23]. This is also our experience at ICDRC. I personally feel that the porphyrin test will revolutionize the way we treat toxic metals in autism. Stay tuned as we are currently involved in several studies looking into this issue further.

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